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L8 ANSWER 4 OF 5 CA COPYRIGHT 2000 ACS
 AN 124:103885 CA
 TI Plasma etching
 IN Nawata, Makoto; Yakushiji, Mamoru
 PA Hitachi Ltd, Japan
 SO Jpn. Kokai Tokkyo Koho, 4 pp.
 CODEN: JKXXAF
 DT Patent
 LA Japanese
 IC ICM H01L021-3065

ICS C23F004-00; H01L021-304
 CC 76-3 (Electric Phenomena)
 FAN.CNT 1

| | PATENT NO. | KIND | DATE | APPLICATION NO. | DATE |
|-------|---|---------------------------------|-------------------------------|-----------------|--------------|
| PI | JP 07263408 | A2 | 19951013 | JP 1994-46817 | 19940317 <-- |
| AB | Plasma etching of Si, Polycryst. Si , or a silicide involves (1) cleaning using a F-contg. gas plasma, (2) seasoning by a gas plasma mixt. contg. .gtoreq.2 selected from HCl, BF3, and/or Cl, (3) pre-etching by plasma of an etching gas, and (4) etching using Cl(g) optionally mixed with O(g). SiF light emission spectra may be monitored for finishing the seasoning and beginning the etching. | | | | |
| ST | silicon plasma etching chlorine gas; silicide | | | | |
| IT | plasma etching chlorine gas | | | | |
| IT | Sputtering | | | | |
| | (etching, plasma etching of (polycryst.) | | | | |
| | silicon or silicide) | | | | |
| IT | Etching | | | | |
| | (sputter, plasma etching of (polycryst.) | | | | |
| | silicon or silicide) | | | | |
| IT | 2551-62-4, Sulfur hexafluoride | 7782-41-4, Fluorine, | uses | 7783-54-2, | |
| | Nitrogen trifluoride | 7790-91-2, Chlorine trifluoride | 13709-36-9, | | |
| Xenon | difluoride | | | | |
| | RL: NUU (Nonbiological use, unclassified); USES (Uses) | | | | |
| | (cleaning gas; plasma etching of (polycryst | | | | |
| | .) silicon or silicide) | | | | |
| IT | 7782-44-7, Oxygen, | uses | 7782-50-5, Chlorine, | uses | |
| | RL: NUU (Nonbiological use, unclassified); USES (Uses) | | | | |
| | (etchant; plasma etching of (polycryst.) | | | | |
| | silicon or silicide) | | | | |
| IT | 11128-24-8P, Silicon fluoride (SiF) | | | | |
| | RL: ANT (Analyte); PNU (Preparation, unclassified); ANST (Analytical | | | | |
| | study); PREP (Preparation) | | | | |
| | (light-emission spectrum; in plasma etching of (| | | | |
| | polycryst.) silicon or silicide) | | | | |
| IT | 7440-21-3, Silicon, | processes | 12651-10-4, Silicide | | |
| | RL: PEP (Physical, engineering or chemical process); PROC (Process) | | | | |
| | (plasma etching of (polycryst.) silicon | | | | |
| | or silicide) | | | | |
| IT | 7637-07-2, Boron trifluoride, | uses | 7647-01-0, Hydrogen chloride, | uses | |
| | RL: NUU (Nonbiological use, unclassified); USES (Uses) | | | | |
| | (seasoning gas before etching; plasma | | | | |
| | etching of (polycryst.) silicon or | | | | |
| | silicide) | | | | |

DERWENT-ACC-NO: 1995-387417
 DERWENT-WEEK: 199550
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TITLE: Plasma etching method for cleaning semiconductor wafer - comprises etching with plasma gas contg. fluorine, treating with mixed gas plasma using e.g. hydrogen chloride and boron tri:fluoride gas, etc..

PRIORITY-DATA:
 1994JP-0046817

March 17, 1994

PATENT-FAMILY:

| PUB-NO | PUB-DATE | LANGUAGE | PAGES | MAIN-IPC |
|---------------|------------------|----------|-------|--------------|
| JP 07263408 A | October 13, 1995 | N/A | 004 | H01L021/3065 |

INT-CL (IPC): C23F 4/00; H01L 21/304; H01L 21/3065

ABSTRACTED-PUB-NO: JP07263408A
 BASIC-ABSTRACT:

Si or polycrystalline-Si or silicide is cleaned by etching using plasma gas contg. F, seasoning treated with mixed gas plasma using HCl gas or BF₃ gas or mixed gas of at least 2 selected from HCl, BF₃, Cl₂ and pre-etching treated by plasma contg. etching gas of Cl or mixed gas of Cl₂ and O₂, and then etched by Cl gas or mixed gas of Cl₂ and O₂.

USE - The method is suitable for cleaning of semiconductor wafer.

ADVANTAGE - Variation of etching speed on Si and oxide film is prevented by restraining effect of remaining F after cleaning.

L9 ANSWER 4 OF 5 JAPIO COPYRIGHT 2000 JPO
 AN 1995-263408 JAPIO
 TI PLASMA ETCHING METHOD
 IN NAWATA MAKOTO; YAKUSHIJI MAMORU
 PA HITACHI LTD, JP (CO 000510)
 PI JP 07263408 A 19951013 Heisei
 AI JP1994-46817 (JP06046817 Heisei) 19940317
 SO PATENT ABSTRACTS OF JAPAN (CD-ROM), Unexamined Applications, Vol. 95, No. 10
 IC ICM (6) H01L021-3065
 ICS (6) C23F004-00; (6) H01L021-304
 AB PURPOSE: To avoid the fluctuation in the residual underneath film for assuring an excellent wafer-to-wafer uniformity by a method wherein the etching step is to be started after cleaning and seasoning steps using HCl, BCl₃ gas plasma as well as preetching step using an etching gas plasma.
 CONSTITUTION: A cleaning gas (SF₆), a seasoning gas (BCl₃) and an etching gas (Cl₂ gas) fed from a feeder 8 by a magnetic field and microwave electric field formed by DC fed from a magnetic field generating DC current 5 to solenoid coils 6, 7 are to be made plasmatic. Next, a preprocessing chamber 4 is cleaned up using cleaning gas (SF₆) plasma while the seasoning and preetching steps of the processing chamber are performed using the seasoning gas (BCl₃ gas plasma) and the etching gas plasma (Cl₂). At this time, a wafer 10 mounted on a mounting electrode 9 is etched away using the etching gas (Cl₂). Through these procedures, the effect of residual fluorine after the cleaning step is averted thereby enabling the fluctuation in the etching rate of Si and an oxide film to be avoided.

L8 ANSWER 4 OF 5 CA COPYRIGHT 2000 ACS
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| IT | 2551-62-4, Sulfur hexafluoride 7782-41-4, Fluorine, uses 7783-54-2, Nitrogen trifluoride 7790-91-2, Chlorine trifluoride 13709-36-9, Xenon difluoride | | | | |
| IT | RL: NUU (Nonbiological use, unclassified); USES (Uses) (cleaning gas; plasma etching of (polycryst .) silicon or silicide) | | | | |
| IT | 7782-44-7, Oxygen, uses 7782-50-5, Chlorine, uses RL: NUU (Nonbiological use, unclassified); USES (Uses) (etchant; plasma etching of (polycryst.) silicon or silicide) | | | | |
| IT | 11128-24-8P, Silicon fluoride (SiF) RL: ANT (Analyte); PNU (Preparation, unclassified); ANST' (Analytical study); PREP (Preparation) (light-emission spectrum; in plasma etching of (polycryst.) silicon or silicide) | | | | |
| IT | 7440-21-3, Silicon, processes 12651-10-4, Silicide RL: PEP (Physical, engineering or chemical process); PROC (Process) (plasma etching of (polycryst.) silicon or silicide) | | | | |
| IT | 7637-07-2, Boron trifluoride, uses 7647-01-0, Hydrogen chloride, uses RL: NUU (Nonbiological use, unclassified); USES (Uses) (seasoning gas before etching; plasma etching of (polycryst.) silicon or silicide) | | | | |

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特開平7-263408

(43) 公開日 平成7年(1995)10月13日

(51) Int.Cl.⁶

識別記号

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F 1

技術表示箇所

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C 2 3 F 4/00

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E 8417-4K

3 4 1 D

H 0 1 L 21/ 302

N

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特願平6-46817

(22) 出願日

平成6年(1994)3月17日

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立製作所機械研究所内

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茨城県土浦市神立町502番地 株式会社日

立製作所機械研究所内

(74) 代理人 弁理士 小川 勝男

(54) 【発明の名称】 プラズマエッチング方法

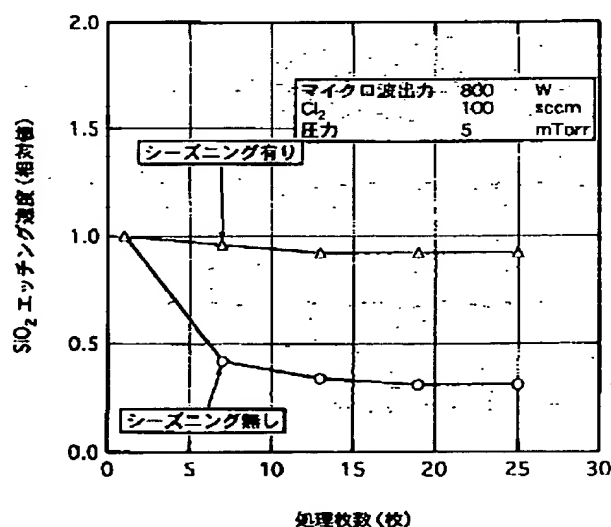
(57) 【要約】

【目的】 クリーニング後のシリコン及び下地膜である酸化膜 (SiO_2) のエッチング速度の変化を抑制しウエハ間の均一性を向上させるのに好適なプラズマエッチング方法を提供することにある。

【構成】 クリーニング後 HCl あるいは BCl_3 ガスプラズマと Cl_2 あるいは Cl_2 と O_2 の混合ガスプラズマでシーズニング並びにプレエッチングを行い、クリーニング後の処理室 (4) 内の残留物の影響を減少させる。

【効果】 クリーニング後の残留フッ素の影響を抑制しシリコン及び酸化膜のエッチング速度の変動を防止することができる。

図 6



【特許請求の範囲】

【請求項1】フッ素を含むガスプラズマによりクリーニングを行い、クリーニング後、塩素ガスの単独ガスあるいは塩素ガスと酸素ガスとの混合ガスをエッチングガスとして用いてシリコン、多結晶シリコン、シリサイドのエッチングを行うエッチング装置において、前記クリーニング後に塩化水素ガス、三弗化硼素ガスの単独ガスあるいは塩化水素ガス、三弗化硼素ガス、塩素ガスの少なくとも2種類以上の混合ガスのプラズマによるシーズニングとエッチングガスのプラズマによるプレエッチングを行った後エッチングを開始することとを特徴とするプラズマエッチング方法。

【請求項2】前記フッ素を含むガスが六フッ化硫黄、三フッ化窒素、二フッ化キセノン、フッ素、三フッ化塩素の単独ガスあるいは混合ガスであることを特徴とする請求項1記載のプラズマエッチング方法。

【請求項3】前記シーズニングにおいてSiFの発光スペクトルをモニターし発光スペクトルの強度の時間変化が一定値以下になった時点でシーズニングを終了しエッチングを開始することとを特徴とする請求項1記載のプラズマエッチング方法。

【発明の詳細な説明】

【0001】

【産業上の利用分野】本発明は、フッ素を含むガスプラズマによりクリーニングを行うものに係り、特に、クリーニング後のシリコン及び下地膜である酸化膜(SiO₂)のエッチング速度の変化を抑制しウエハ間の均一性を向上させるのに好適なプラズマエッチング方法に関するものである。

【0002】

【従来の技術】従来、エッチングを含めたプラズマプロセスではウエハの粒子汚染を防止するためにクリーニングを行いクリーニング後の処理室内の残留物をなくすためにポストクリーニングを行っている。SF₆、NF₃ガスをクリーニングに用いた場合にはN₂、Ar、H₂、O₂ガスプラズマがポストクリーニングに用いられている。

【0003】なお、本技術に関するものとして、例えば、文献：平塚豊著、洗浄設計P41-53、1992. Summerが挙げられる。

【0004】

【発明が解決しようとする課題】従来のエッチング方法では、クリーニング後の処理室内の残留物のエッチング特性に及ぼす影響について考慮されておらず、クリーニング後処理枚数とともにシリコン及び下地膜の酸化膜のエッチング速度が減少し、下地酸化膜の残膜が変動するという問題点があった。

【0005】本発明は、クリーニング後のシリコン及び酸化膜のエッチング速度の減少を抑制し下地酸化膜の残膜の変動を防止し良好なウエハ間の均一性が得られるエ

ッチング方法を提供することにある。

【0006】

【課題を解決するための手段】上記目的を解決するために、クリーニング後HCl、BCl₃ガスプラズマによるシーズニングとエッチングガスであるCl₂あるいはCl₂とO₂の混合ガスプラズマによるプレエッチングを行い、クリーニング後の処理室内の残留物の影響を減少させようとしたものである。

【0007】

10 【作用】図1に、SF₆ガスプラズマでクリーニングを行った後、Cl₂ガスプラズマでシリコンをエッチングした場合におけるSiF(波長441nm)の発光スペクトルの処理枚数による変化を示す。シリコンとフッ素の反応によって生成するSiFの発光スペクトルの強度は処理枚数とともに減少しほぼ一定となる。このことからフッ素を含むガスによるクリーニング後、処理室内にはフッ素が残留していることが分かった。図2、図3に、Cl₂ガスにSF₆ガスを添加した場合のSiFの発光スペクトルとシリコン及び酸化膜のエッチング速度の変化を示す。図2、図3に示すようにSF₆の添加量の増加とともにシリコン及び酸化膜のエッチング速度は増加する。また、SF₆の添加量の増加とともにSiF(波長441nm)の発光スペクトルの強度は増加する。このことから残留フッ素によりシリコン及び酸化膜のエッチング速度は変動し、残留フッ素の減少とともにシリコン及び酸化膜のエッチング速度が低下することを見出した。したがって、クリーニングの後残留フッ素の除去のためHCl、BCl₃ガスプラズマによるシーズ

20 30 40 50
ニングを行い、SiFの発光スペクトルの強度の時間変化が一定値以下になった時点でシーズニングを終了し、シーズニング後エッチングガスであるCl₂あるいはCl₂とO₂の混合ガスプラズマでプレエッチングを行う。シーズニング及びプレエッチングの後、エッチングを開始することによりシリコン及び酸化膜のエッチング速度の変動を抑制できる。また、図4に示すように、Clに比べFとの結合エネルギーが大きいHあるいはBを含むHCl、BCl₃ガスプラズマを用いることにより残留フッ素の除去時間を短縮できることを見出した。

【0008】

【実施例】本発明の一実施例を図5により説明する。図5は、マイクロ波プラズマエッチング装置の概略図を示したものである。図5において、マグネトロン1から発振したマイクロ波は導波管2を伝播しマイクロ波導入窓3を介して処理室4に導かれる。磁界発生用直流電源5からソレノイドコイル6、7に供給される直流電流によって形成される磁界とマイクロ波電界によってエッチングガス供給装置8から供給されるクリーニングガス(SF₆)、シーズニングガス(BCl₃)及びエッチングガス(Cl₂ガス)はプラズマ化される。SF₆ガスプラズマにより処理室4のクリーニングが行われる。処理室4

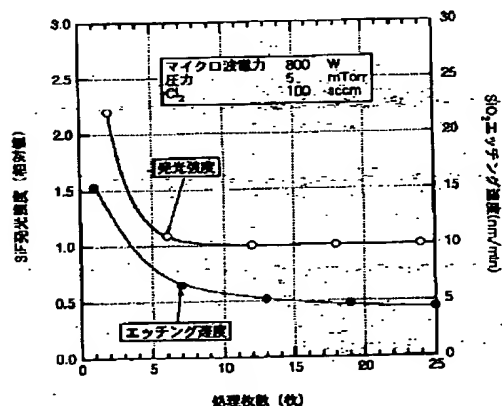
のシーズニング及びアレエッチングが、 BCl_3 ガスプラズマと Cl_2 ガスプラズマにより行われる。 Cl_2 ガスにより載置電極9に載置されているウエハ10がエッチングされる。クリーニング、エッチング時の圧力は真空排気装置11によって制御される。また、ウエハに入射するイオンのエネルギーは載置電極9に高周波電源12から供給される高周波電力によって制御される。図6、図7にシーズニングの有無によるシリコン及び酸化膜のエッチング速度の変化の違いを示す。シーズニング及びアレエッチングは BCl_3 ガスプラズマと Cl_2 ガスプラズマにより行い、SiFの発光スペクトルを10秒毎にモニタし時間 t_n と時間 t_{n-1} に測定したスペクトルの発光強度比が 1 ± 0.002 になった時点でシーズニングを停止した。クリーニング後にシーズニングを行うことによりクリーニング時に生成されるフッ素の残留の影響を抑制しエッチング速度の変動を防止できる。

【0009】本一実施例によれば、クリーニング後の残留フッ素の影響を抑制しシリコン及び酸化膜のエッチング速度の変動を防止することができる。

【0010】本一実施例ではマイクロ波プラズマエッチング装置についてその効果を説明したが、他の放電方式例えば反応性イオンエッチング(RIE)、プラズマモードエッチング(PE)、マグネトロンRIE、ヘリコン、TCPにおいても同様な効果が得られる。

【図1】

図 1



【0011】

【発明の効果】本発明によれば、クリーニング後の残留フッ素の影響を抑制しシリコン及び酸化膜のエッチング速度の変動を防止することができる。

【図面の簡単な説明】

【図1】SiF発光強度の処理枚数依存性示す説明図である。

【図2】SiF発光強度の SF_6 添加量依存性を示す説明図である。

【図3】Si及び SiO_2 エッチング速度の SF_6 添加量依存性を示す説明図である。

【図4】SiF発光強度の処理時間依存性示す説明図である。

【図5】マイクロ波プラズマエッチング装置の構成図である。

【図6】本発明の一実施例での効果を説明するための SiO_2 エッチング速度の処理枚数依存性示す説明図である。

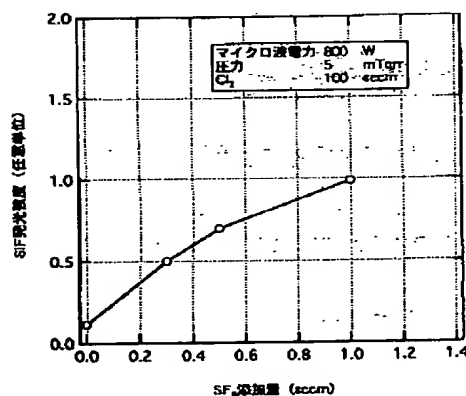
【図7】本発明の一実施例での効果を説明するためのSiエッチング速度の処理枚数依存性示す説明図である。

【符号の説明】

2…導波管、3…マイクロ波導入窓、4…処理室、6…ソレノイドコイル。

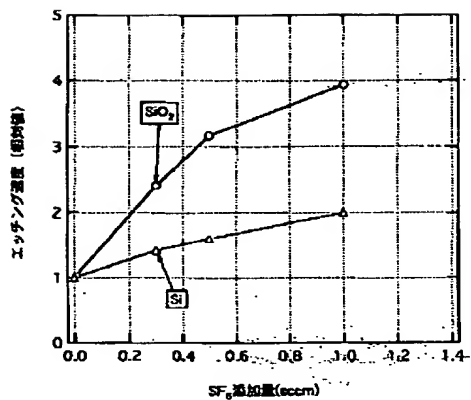
【図2】

図 2



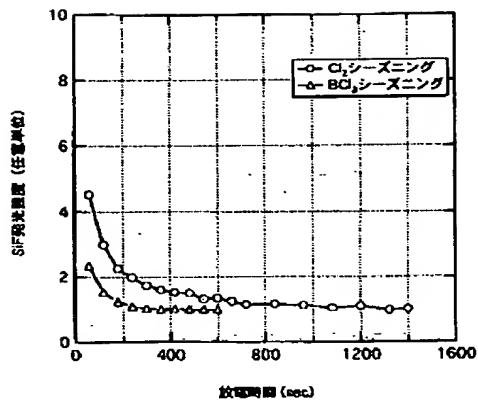
【図3】

図 3



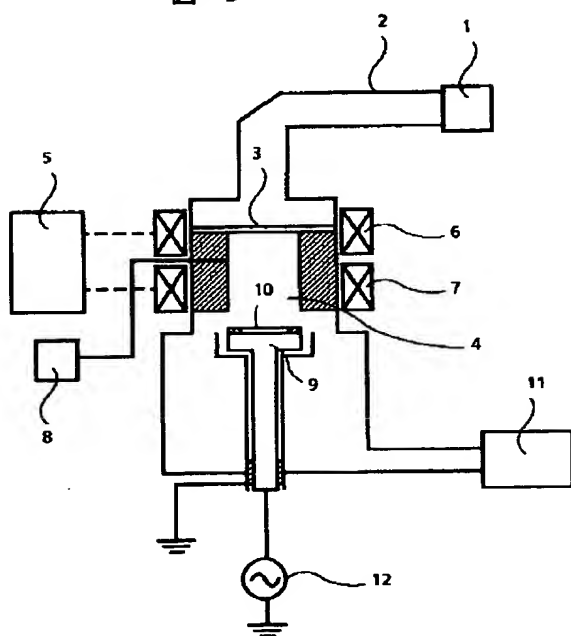
【図4】

図 4



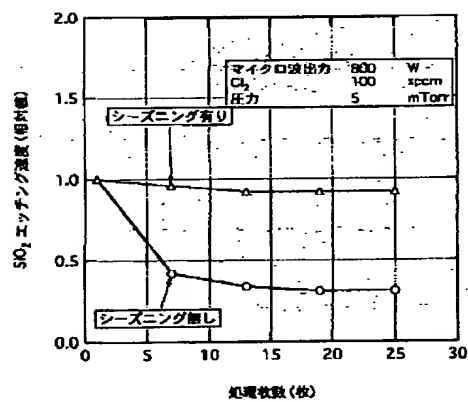
【図5】

図 5



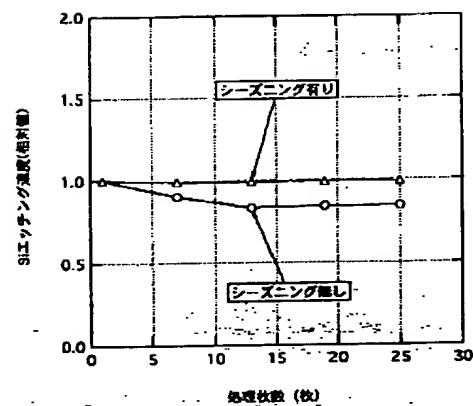
【図6】

図 6



【図7】

図 7



07-263408

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CLAIMS

[Claim(s)]

[Claim 1] It cleans by the gas plasma containing a fluorine. After cleaning, In the etching system which performs etching of silicon, polycrystal silicon, and a silicide, using the mixed gas of the independent gas of chlorine gas or chlorine gas, and oxygen gas as etching gas After the aforementioned cleaning, hydrogen chloride gas, Starting etching, after performing seasoning by the plasma of at least two or more kinds of mixed gas of the independent gas of boron-trifluoride gas or hydrogen chloride gas, boron-trifluoride gas, and chlorine gas, and pre-etching by the plasma of etching gas The plasma-etching technique by which it is characterized.

[Claim 2] The plasma-etching technique according to claim 1 characterized by the gas containing the aforementioned fluorine being the independent gas or mixed gas of 2 3 fluoride [6 fluoride / sulfur / and nitrogen] and xenon fluoride, a fluorine, and 3 fluoride-salt **.

[Claim 3] The plasma-etching technique according to claim 1 characterized by ending seasoning and starting etching when it acts as the monitor of the emission spectrum of SiF in the aforementioned seasoning and time change of the intensity of an emission spectrum becomes below a constant value.

[Translation done.]

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DETAILED DESCRIPTION

[Detailed Description of the Invention]

[0001]

[Field of the Invention] It is related with the suitable plasma-etching technique for this invention to relate to what cleans by the gas plasma containing a fluorine, suppress especially change of the etch rate of the oxide film (SiO₂) which is the silicon and the substratum layer after cleaning, and raise the homogeneity between wafers.

[0002]

[Description of the Prior Art] Conventionally, in the plasma process including etching, in order to clean in order to prevent grain contamination of a wafer, and to lose the residue of the processing interior of a room after cleaning, post cleaning is performed. When SF₆ and NF₃ gas are used for cleaning, N₂, Ar, H₂, and O₂ gas plasma are used for post cleaning.

[0003] In addition, reference: Hiratsuka ****, washing design P 41-53, and 1992. Summer are mentioned as a thing about this technique.

[0004]

[Problem(s) to be Solved by the Invention] By the conventional etching technique, it was not taken into consideration about the influence affect the etching property of the residue of the processing interior of a room after cleaning, but the etch rate of the oxide film of silicon and a substratum layer decreased with cleaning after-treatment number of sheets, and there was a trouble of changing **** of a substratum oxide film.

[0005] this invention is to offer the etching technique by which a decrement of the silicon after cleaning and the etch rate of an oxide film is suppressed, change of **** of a substratum oxide film is prevented, and the homogeneity between good wafers is acquired.

[0006]

[Means for Solving the Problem] In order to solve the above-mentioned purpose, pre-etching by after [cleaning] HCl, Cl₂ which is seasoning and etching gas by BCl₃ gas plasma, or the mixed-gas plasma of Cl₂ and O₂ tends to be performed, and it is going to decrease the influence of the residue of the processing interior of a room after cleaning.

[0007]

[Function] After cleaning to drawing 1 with SF₆ gas plasma, change by the processing number of sheets of the emission spectrum of SiF (wavelength of 441nm) at the time of etching silicon with Cl₂ gas plasma is shown in it. The intensity of the emission spectrum of SiF generated by the reaction of silicon and a fluorine decreases with processing number of sheets, and becomes almost fixed. The processing interior of a room found that the fluorine remained after cleaning by the gas which contains a fluorine from this. Change of the etch rate of the emission spectrum of SiF at the time of adding SF₆ gas in Cl₂ gas, silicon, and an oxide film is shown in drawing 2 and the drawing 3. As shown in drawing 2 and the drawing 3, the etch rate of silicon and an oxide film increases with the increase in the addition of SF₆. Moreover, the intensity of the emission spectrum of SiF (wavelength of 441nm) increases with the increase in the addition of SF₆. The etch rate of silicon and an oxide film was changed by the remains fluorine from this, and it found out that the etch rate of silicon and an oxide film fell with a decrement of a remains fluorine. Therefore, seasoning by HCl and BCl₃ gas plasma is performed for elimination of the post-remains fluorine of cleaning, when time change of the intensity of the emission spectrum of SiF becomes below a constant value, seasoning is ended, and pre-etching is performed with Cl₂ or the mixed-gas plasma of Cl₂ and O₂ which is after [seasoning] etching gas. Change of the etch rate of silicon and an oxide film can be suppressed by starting etching after seasoning and pre-etching. Moreover, as shown in drawing 4, it found out that the elimination time of a remains fluorine could be shortened by using HCl and BCl₃ gas plasma which contain H or B with large binding energy with F compared with Cl.

[0008]

[Example] Drawing 5 explains one example of this invention. Drawing 5 shows the schematic diagram of a microwave plasma etching system. In drawing 5, the microwave oscillated from the magnetron 1 spreads a waveguide 2, and is led to the processing room 4 through the microwave introduction aperture 3. The cleaning gas (SF₆), the seasoning gas (BCl₃), and etching gas (Cl₂ gas) which are supplied by the magnetic field formed of the direct current supplied to solenoid coils 6 and 7 from DC power supply for magnetic-field occurrence 5 and the microwave electric field from the etching gas supply system 8 are plasma-ized. Cleaning of the processing room 4 is performed by SF₆ gas plasma. Seasoning and pre-etching of the processing room 4 are performed by BCl₃ gas plasma and Cl₂ gas plasma. It is etched in the wafer 10 currently laid in the installation electrode 9 by Cl₂ gas. The pressure at the time of cleaning and etching is controlled by the evacuation equipment 11. Moreover, the energy of the

ion which carries out incidence to a wafer is controlled by RF power supplied to the installation electrode 9 from RF generator 12. The difference in change of the etch rate of the silicon by the existence of seasoning and an oxide film is shown in drawing 6 and the drawing 7. BCl₃ gas plasma and Cl₂ gas plasma performed seasoning and pre-etching, and when the photogenesis intensity ratio of the spectrum which carried out the monitor of the emission spectrum of SiF every 10 seconds, and was measured to time t_n and time t_{n-1} was set to 1×0.002 , they suspended seasoning. By performing seasoning after cleaning, the influence of remains of the fluorine generated at the time of cleaning is suppressed, and change of an etch rate can be prevented.

[0009] According to this 1 example, the influence of the remains fluorine after cleaning can be suppressed, and change of the etch rate of silicon and an oxide film can be prevented.

[0010] Although this 1 example explained the effect about the microwave plasma etching system, the same effect is acquired also in other electric discharge methods (RIE), for example, reactive ion etching, and plasma mode etching (PE) magnetron RIE, Helicon, and TCP.

[0011]

[Effect of the Invention] According to this invention, the influence of the remains fluorine after cleaning can be suppressed, and change of the etch rate of silicon and an oxide film can be prevented.

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[Function] After cleaning to drawing 1 with SF₆ gas plasma, change by the processing number of sheets of the emission spectrum of SiF (wavelength of 441nm) at the time of etching silicon with Cl₂ gas plasma is shown in it. The intensity of the emission spectrum of SiF generated by the reaction of silicon and a fluorine decreases with processing number of sheets, and becomes almost fixed. The processing interior of a room found that the fluorine remained after cleaning by the gas which contains a fluorine from this. Change of the etch rate of the emission spectrum of SiF at the time of adding SF₆ gas in Cl₂ gas, silicon, and an oxide film is shown in drawing 2 and the drawing 3. As shown in drawing 2 and the drawing 3, the etch rate of silicon and an oxide film increases with the increase in the addition of SF₆. Moreover, the intensity of the emission spectrum of SiF (wavelength of 441nm) increases with the increase in the addition of SF₆. The etch rate of silicon and an oxide film was changed by the remains fluorine from this, and it found out that the etch rate of silicon and an oxide film fell with a decrement of a remains fluorine. Therefore, seasoning by HCl and BCl₃ gas plasma is performed for elimination of the post-remains fluorine of cleaning, when time change of the intensity of the emission spectrum of SiF becomes below a constant value, seasoning is ended, and pre-etching is performed with Cl₂ or the mixed-gas plasma of Cl₂ and O₂ which is after [seasoning] etching gas. Change of the etch rate of silicon and an oxide film can be suppressed by starting etching after seasoning and pre-etching. Moreover, as shown in drawing 4, it found out that the elimination time of a remains fluorine could be shortened by using HCl and BCl₃ gas plasma which contain H or B with large binding energy with F compared with Cl.

[0008]

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Field

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Technique

[Description of the Prior Art] Conventionally, in the plasma process including etching, in order to clean in order to prevent grain contamination of a wafer, and to lose the residue of the processing interior of a room after cleaning, post cleaning is performed. When SF6 and NF3 gas are used for cleaning, N2, Ar, H2, and O2 gas plasma are used for post cleaning.

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Effect

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TECHNICAL PROBLEM

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MEANS

[Means for Solving the Problem] In order to solve the above-mentioned purpose, pre-etching by after [cleaning] HCl, Cl₂ which is seasoning and etching gas by BCl₃ gas plasma, or the mixed-gas plasma of Cl₂ and O₂ tends to be performed, and it is going to decrease the influence of the residue of the processing interior of a room after cleaning.

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OPERATION

[Function] After cleaning to drawing 1 with SF6 gas plasma, change by the processing number of sheets of the emission spectrum of SiF (wavelength of 441nm) at the time of etching silicon with Cl2 gas plasma is shown in it. The intensity of the emission spectrum of SiF generated by the reaction of silicon and a fluorine decreases with processing number of sheets, and becomes almost fixed. The processing interior of a room found that the fluorine remained after cleaning by the gas which contains a fluorine from this. Change of the etch rate of the emission spectrum of SiF at the time of adding SF6 gas in Cl2 gas, silicon, and an oxide film is shown in drawing 2 and the drawing 3. As shown in drawing 2 and the drawing 3, the etch rate of silicon and an oxide film increases with the increase in the addition of SF6. Moreover, the intensity of the emission spectrum of SiF (wavelength of 441nm) increases with the increase in the addition of SF6. The etch rate of silicon and an oxide film was changed by the remains fluorine from this, and it found out that the etch rate of silicon and an oxide film fell with a decrement of a remains fluorine. Therefore, seasoning by HCl and BCl3 gas plasma is performed for elimination of the post-remains fluorine of cleaning, when time change of the intensity of the emission spectrum of SiF becomes below a constant value, seasoning is ended, and pre-etching is performed with Cl2 or the mixed-gas plasma of Cl2 and O2 which is after [seasoning] etching gas. Change of the etch rate of silicon and an oxide film can be suppressed by starting etching after seasoning and pre-etching. Moreover, as shown in drawing 4, it found out that the elimination time of a remains fluorine could be shortened by using HCl and BCl3 gas plasma which contain H or B with large binding energy with F compared with Cl.

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EXAMPLE

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DESCRIPTION OF DRAWINGS

[Brief Description of the Drawings]

[Drawing 1] It is processing number-of-sheets dependency **** explanatory drawing of SiF photogenesis intensity.

[Drawing 2] It is explanatory drawing showing SF6 addition dependency of SiF photogenesis intensity.

[Drawing 3] It is explanatory drawing showing SF6 addition dependency of Si and SiO2 etch rate.

[Drawing 4] It is processing-time dependency **** explanatory drawing of SiF photogenesis intensity.

[Drawing 5] It is the block diagram of a microwave plasma etching system.

[Drawing 6] It is processing number-of-sheets dependency **** explanatory drawing of SiO2 etch rate for explaining the effect in one example of this invention.

[Drawing 7] It is processing number-of-sheets dependency **** explanatory drawing of Si etch rate for explaining the effect in one example of this invention.

[Description of Notations]

2 [-- A processing room, 6 / -- Solenoid coil.] -- A waveguide, 3 -- A microwave introduction aperture, 4

[Translation done.]